## **Mass Spectrometry**

## **Question Paper**

Level	Pre U	
Subject	Chemistry	
Exam Board	Cambridge International Examinations	
Topic	Mass Spectrometry	
Booklet	Question Paper	

Time Allowed: 12 minutes

Score: /10

Percentage: /100

**Grade Boundaries:** 

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ı. <sup>-</sup>	The molecule shown is but-3-ynoic acid.		
		ОН	
(6	a)	Give the molecular formula for but-3-ynoic acid.	
		molecular formula[1]	
(1	b)	Draw the structure and name an isomer of but-3-ynoic acid that contains the same functional groups.	
		structure	
		name[2]	
(	c)	Work out the percentage composition (by mass) of the constituent elements in but-3-ynoic acid.	
		C% H% O% [2]	
(	d)	Give the m/z value of the molecular ion peak in the mass spectrum of but-3-ynoic acid.	

.....

[1]

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(e)	(i)	But-3-ynoic acid contains a carboxylic acid group with double and single bonds. Write down a value in cm <sup>-1</sup> that falls in the wavenumber range of each of these types of bonds in an infrared spectrum.				
		double bondscm <sup>-1</sup>				
		single bonds (not involving hydrogen) .	cm <sup>-1</sup>			
		single bonds to hydrogen .	cm <sup>-1</sup> [2	2]		
	(ii)	ii) The carboxylic acid O–H stretch has a characteristic appearance in an infrared spectrum. Describe its general appearance. There is no need to give wavenumbe values.				
			[1	]		
(f)	(f) Scientists recently isolated a novel, highly toxic and unstable molecule, T, fro poisonous Asian mushroom Russula subnigricans (reported in Nature Che Biology, 2009).					
	<b>T</b> is an isomer of but-3-ynoic acid. Its infrared spectrum indicates that <b>T</b> also contains carboxylic acid group. Its carbon-13 nmr spectrum, however, only contains 3 signals.					
	Sug	ggest a structure for <b>T</b> .				

[1]

[Total: 10]